Phys/Chem Weekly Planner: All science week of 2.24.2020



Objectives for the week: Chm.2.2 Analyze the structure and nature of chemical quantities. Phys 1.2 Analyze the nature of circular motion.

Дау	Honors Physics	Honors Chemistry
Mon	TEST from 2.20.2020	TEST from 2.20.2020
2/24	*HW= Test corrections,	
,	Read/take notes on ch. 6.2 and show work to solve this	*HW= <mark>Test corrections, Read</mark> Ch 9
	problem:	Complete worksheet 9a
	Uniform Circular Motion A 13-g rubber stopper is attached	Worksheet 9a balancing
	string. The stopper is swung in a horizontal circle, making one	re
	in 1.18 s. Find the tension force exerted by the string on the s	to
Tues 2/25	ACT	ACT
	*HW= Test corrections, Read/take notes on ch. 6.2 and show work to solve this problem:	*HW= TEST corrections, Read/take notes on Ch 9
	Uniform Circular Motion A 13-g rubber stopper is attached	tt <mark>Worksheet 9.1 balancing</mark>
	string. The stopper is swung in a horizontal circle, making one re in 1.18 s. Find the tension force exerted by the string on the sto	
Wed	-Questions from TEST?	-Questions from test?
2/26	-Circular motion notes	-Go over balance
Progr	Circular motion homework	equations homework
ess	A	Delensing chamical equations
repor tc		Balancing chemical equations
<mark>ts</mark>		
	*HW= Test corrections due	*HW= Test corrections due Thursday.

Thurs 2/27	Thursday. Circular motion homework A due Thursday. HAND IN HW and corrections *HW= Circular motion problems #192-207 circular motion problems :)	Hand in HW https://www.youtube.com/wat ch?v=AkWYrp8Z5X4 Take NOTES on above WORD equations and types of reactions Types of reactions and WORD equations $C_5H_{12} + 8O_2 \rightarrow 5CO_2 + 6H_2O$ Pentane burns in oxygen gas to form carbon dioxide and water. Combustion reaction *HW=1-8 and 1-8 and 1-8 due FRII
Friday 2/28	QUIZ- F _c CENTRIPETAL FORCE LAB!! *HW=DUE Mon: 186-207	QUIZ- equations LAB GAMES DAY!!!!!!! *HW= Due MON, do 3 problems from each of the 3 pages of worksheet.
	C_2H_{e}	$_{5} + O_{2} \rightarrow CO_{2} + H_{2}O$ $_{5} + 3.5 O_{2} \rightarrow 2 CO_{2} + 3H_{2}O$ $H_{6} + 7O_{2} \rightarrow 4CO_{2} + 6H_{2}O$

Warm up activities! This is RARE but we

have to cancel warm ups for this week and plan on

exit passes only.



Chem worksheet 9a:

1. Answer the following questions about the chemical equation shown below:

 $2 H_2 + O_2 \rightarrow 2 H_2O$

- a) What are the reactants?
- b) What is the product?
- c) What do we call the number "2" in front of the H_2 (and H_2O)?
- d) Is the reaction balanced?
- e) Why is there not a coefficient for O_2 ?.
- f) How many hydrogen atoms are needed to produce two H₂O molecules?
- g) How many oxygen atoms are needed to produce two H₂O molecules?
- h) How many hydrogen molecules are needed to produce two H₂O molecules?
- i) How many oxygen molecules are needed to produce two H₂O molecules?
- i) Write the "word equation" that you would use to describe this reaction.
 - Use words in a sentence, not formulas or an arrow. *Ignore the coefficients*.
- 2. Balance the following chemical equations:
- a) Fe + O₂ \rightarrow Fe₂O₃
- b) $H_2 + Cl_2 \rightarrow HCl$
- c) Ag + H₂S \rightarrow Ag₂S + H₂
- d) $CH_4 + O_2 \rightarrow CO_2 + H_2O$
- e) HgO \rightarrow Hg + O₂
- f) Co + H₂O \rightarrow Co₂O₃ + H₂

Monday 2.24.20- <u>https://evansccca.weebly.com/</u>

TURN OFF cell phone and put in the bin (\mathfrak{O})

	CHEM Warm up:
cell phone and put in the bin	Turn OFF your cell phone and
	put in bin 🚱

Draw a picture of any physics problem.	Draw a picture of any chemistry problem.
<u> Tuesday 2.25.20-</u>	https://evansccca.weebly.com/
PHYZ Warm up: TURN OFF cell phone and put in the bin ③	CHEM Warm up: Turn OFF your cell phone and put in bin 🐵

Wednesday 2.26.20- https://evansccca.weebly.com/

PHYZ Warm up: TURN OFF cell phone and put in the bin	CHEM Warm up: Turn OFF your cell phone and put in bin 🐵	
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Thursday 2.27.20- https://evansccca.weebly.com/

	CHEM Warm up: Turn OFF your cell phone and put in bin 😳	
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Friday 2.28.20- https://evansccca.weebly.com/

cell phone and put in the bin	CHEM Warm up: Turn OFF your cell phone and put in bin 🕹